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A convenient method for the preparation of chiral phosphonoacetamides and their Horner–Wadsworth–Emmons reaction

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Abstract—Chiral phosphonoacetamides bearing (*S*)-(α -methylbenzyl)benzylamine, (*S*,*S*)-bis(α -methylbenzyl)amine, L-phenylglycine methyl ester and L-phenylglycinol were easily prepared in good yield by means of the Michaelis–Arbuzov reaction of chiral bromoacetamides obtained in quantitative yield, with trimethyl phosphite, which under Horner–Wadsworth–Emmons conditions with several aryl and alkyl aldehydes under Masamune–Roush procedure using LiCl and DBU in THF or toluene gave the corresponding chiral α , β -unsaturated amides. The present procedure is a convenient and efficient methodology for the preparation of phosphonoacetamides and chiral α , β -unsaturated amides in high *E*-selectivity.

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1. Introduction

The stereodefined synthesis of carbon-carbon double bonds with high selectivity is critically important in organic synthesis, since geometrically-defined olefins serve as good building blocks in organic synthesis.¹ Of particular utility are the olefins bearing electron-withdrawing groups, such as the carbonyl group, since these groups are not only viable for further functionalization, but they also activate the olefin moiety for reactions such as Michael addition² and cycloaddition reactions.³ Additionally, α,β -unsaturated amides belong to an important class of natural products. which show both biological⁴ and insecticide activities.⁵ For these reasons, the preparation of α,β -unsaturated amides is a topic of great interest, in this context the Wittig reaction and related reactions have served as the most powerful method for the construction of double bonds.⁶ The Horner-Wadsworth-Emmons (HWE) modification of the Wittig reaction in particular is widely employed.^{7,8} The

reactions of aldehydes with phosphonates bearing α -substituents that stabilize the carbanion preferentially furnish the corresponding *E*-alkenes. Recently, we described the highly selective synthesis of chiral (*E*)- α , β -unsaturated amides bearing (*S*)- α -methylbenzylamine from phosphonoacetamides via a HWE reaction.⁹ As an extension of this methodology, in this work we herein report the synthesis of several chiral α , β -unsaturated amides bearing (*S*)-(α -methylbenzyl)benzylamine,¹⁰ (*S*,*S*)-bis(α -methylbenzyl)amine,¹¹ (*S*)-phenylglycine methyl ester¹² and (*S*)-phenylglycinol¹³ via a HWE reaction from the corresponding phosphonoacetamides **4–8**.

2. Results and discussion

The starting chiral phosphonoacetamides **4–6** were easily prepared by employing the protocol recently developed in our laboratories.¹⁴ In this context, the treatment of the appropriate chiral amine¹⁵ with bromoacetyl bromide in the presence of K_2CO_3 in a CH₂Cl₂/H₂O mixture at room temperature, afforded the corresponding chiral α -bromo-

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acetamides 1–3 in excellent yields,¹⁶ which by means of the Michaelis–Arbuzov reaction¹⁷ with trimethyl phosphite gave the chiral phosphonoacetamides **4–6** in 92–98% yield (Scheme 1).¹⁸



Scheme 1.

On the other hand, reduction of the methyl ester group in phosphonoacetamide **6** with sodium borohydride in THF/ MeOH at reflux¹⁹ produced phosphonoacetamide **7** in 89% yield. Additionally, treatment of phosphonoacetic acid **9**, obtained by hydrolysis from the corresponding ethyl ester derivative, with L-phenylglycinol²⁰ in the presence of dicyclohexylcarbodiimide (DCC) and catalytic amounts of 4-(*N*,*N*-dimethylamino)pyridine (DMAP) in dichloromethane at room temperature, afforded phosphonoacetamide **7** in 97% yield. In a similar manner, **8** was obtained in 93% yield from the corresponding phosphonoacetic acid **10** (Scheme 2).²¹



Scheme 2.

With phosphonoacetamides **4–8** in hand, we initiated our own investigation by examining the HWE reaction of **4–8** with several aryl and alkyl aldehydes to obtain the corresponding chiral α,β -unsaturated amides. Thus, in the first instance and following the Masamune–Roush procedure,²² phosphonoacetamide **4** in dry THF was treated with 8-diazabicyclo[5.4.0]undec-7-ene in the presence of LiCl, followed by the addition of benzaldehyde at room temperature.²³ The reaction was completed within 2 h, and a >98:02 ratio²⁴ of *E:Z* α,β -unsaturated amide **11a** was obtained in 93% yield (Table 1, entry 1). Identical results were obtained in the reaction of 4 with *p*-chlorobenzaldehyde, *p*-methoxybenzaldehyde and trimethylacetaldehyde (Table 1, entries 2–4). However, only moderate selectivity E:Z = 87:13 was obtained with isobutyraldehyde and isovaleraldehyde (Table 1, entries 5 and 6). These results suggest that the stereoselectivity in the HWE reaction of 4 with aromatic and alkyl aldehydes is determined by the steric hindrance at the α -position in the corresponding aldehydes.

Table 1. HWE reaction of 4 with several aldehydes

4	RCHO DBU/LiCI THF, r.t.	R N Ph Bn (E)-11a-f	+ 0 N Bn (Z)-11a-f	le ∑Ph
Entry		R	Yield ^a (%)	$E:Z^{\mathbf{b}}$
1	a	C ₆ H ₅	93	98:02
2	b	p-Cl–C ₆ H ₄	94	98:02
3	с	p-MeO-C ₆ H ₄	97	98:02
4	d	t-Bu	92	98:02
5	e	<i>i</i> -Pr	94	83:17
6	f	<i>i</i> -Bu	95	83:17
7	e	<i>i</i> -Pr	92	96:04 ^c
8	f	<i>i</i> -Bu	90	93:07 ^c

^a Isolated yield after purification by column chromatography.

^b The *E*:*Z* ratio was determined by ¹H NMR on crude product on the basis of the olefinic protons.

^c The reaction was carried out in toluene at 100 °C.

On the other hand, it is generally accepted that the stereoselectivity in the HWE reactions is a result of both kinetic and thermodynamic control upon the reversible formation of the *erythro* and *threo* adducts and their decomposition to olefins.⁷ Therefore, the stereochemistry is determined by a combination of the stereoselectivity in the initial carbon–carbon bond-forming step and reversibility of the intermediate adducts. The predominant formation of the *E*-olefins in the case of dialkylphosphono derivatives can be explained by the formation of thermodynamically more stable *threo* adducts. In this context, when we carried out the HWE reaction of **4** with isobutyraldehyde and isovaleraldehyde in toluene at 100 °C, a good *E*-selectivity and yield of α , β -unsaturated amides **11e–f** were obtained (Table 1, entries 7 and 8).²⁵

In a similar way, the reaction of phosphonoacetamide **5** afforded the chiral α , β -unsaturated amides **12a**–**f** with excellent *E*-selectivity and yield. The results are summarized in Table 2.²⁶

On the other hand, the HWE reaction of **6** with benzaldehyde, *p*-chlorobenzaldehyde, *p*-methoxybenzaldehyde and trimethylacetaldehyde at room temperature provided the (S)-(E)- α , β -unsaturated amides **13a**-**d** as the only product (Table 3, entries 1–4).²⁷ However, the reaction of **6** with isobutyraldehyde and isovaleraldehyde under the same conditions, gave the α , β -unsaturated amides **13e**-**f** with low selectivities *E*:*Z* (Table 3, entries 5 and 6), whereas at

Table 2. HWE reaction of 5 with several aldehydes



^a Isolated yield after purification by column chromatography.

^b The E:Z ratio was determined by ¹H NMR on the crude product.

^c The reaction was carried out in toluene at 100 °C.

50 °C the chiral α , β -unsaturated amides **13e–f** were obtained with better *E*-selectivities (Table 3, entries 7 and 8).

Table 3. HWE reaction of 6 with several aldehydes

6 - [RCHO DBU/LiCI THF, r.t.	R Ph H O (<i>E</i>)-13a-f		Ph O OMe
Entry		R	Yield ^a (%)	$(E):(Z)^{b}$
1	a	C ₆ H ₅	80	98:02
2	b	p-Cl–C ₆ H ₄	90	98:02
3	c	p-MeO-C ₆ H ₄	90	98:02
4	d	t-Bu	70	98:02
5	e	<i>i</i> -Pr	88	47:53
6	f	<i>i</i> -Bu	90	65:35
7	e	<i>i</i> -Pr	90	62:38 ^c
8	f	<i>i</i> -Bu	91	83:17 ^c

^a Isolated yield after purification by column chromatography.

^b The *E*:*Z* ratio was determined by ¹H NMR on the crude product.

^c The reaction was carried out at 50 °C.

Finally, an excellent E-selectivity and good yield were obtained in the reaction of phosphonoacetamide 8 with benzaldehyde, p-chlorobenzaldehyde, p-methoxybenzaldehyde, trimethylacetaldehyde and isovaleraldehyde (Table 4, entries 1–5), where only the (S)-(E)- α , β -unsaturated amides 14a–e were detected by ¹H NMR, whereas the reaction of 8 with isobutyraldehyde afforded the α , β -unsaturated amides 14f with a selectivity E: Z = 90:10 and 89% (Table 4, entry 5). On the other hand, the reaction of 7 with isobutyraldehyde gave the α , β -unsaturated amides **14f** with a selectivity E:Z = 70:30 and 97% yield (Table 4, entry 7), whereas the reaction with isovaleraldehyde produced the chiral α,β unsaturated amides 14e with a selectivity E:Z = 94:06and 88% yield (Table 4, entry 8). These results show that diethylphosphonoacetamide 8 gave an E-selectivity higher than dimethylphosphonoacetamide 7.

No loss in the enantiomeric purity for phosphonoacetamides **6** and **8** was observed in the process by control experiTable 4. HWE reaction of 7 and 8 with several aldehydes



^a Isolated yield after purification by column chromatography.

^b The *E*:*Z* ratio was determined by ¹H NMR in the crude product.

^c The reaction was carried out using the phosphonoacetamide 7.

ment, in which **6** and **8** were submitted to the reaction conditions in the absence of the aldehyde at room temperature and 50 °C, recovering the corresponding phosphonoacetamides **6** and **8** without any loss in specific rotation.

3. Conclusions

In conclusion, we have developed a practical and efficient methodology for the synthesis of chiral phosphonoacetamides **4–8**, and chiral α,β -unsaturated amides **11–14** in excellent *E*-selectivity under thermodynamic control. These chiral (*E*)- α,β -unsaturated amides should be excellent building blocks in organic synthesis.

4. Experimental

Optical rotations were taken on a Perkin-Elmer 241 polarimeter in a 1 dm tube; concentrations are given in g/100 mL. For the flash chromatography, Silica Gel 60 (230-400 mesh ASTM) was used. ¹H NMR spectra were registered on a Varian Mercury 200 and 300 MHz, INOVA 400 (400 MHz) and ¹³C NMR on a Varian Mercury 50 and 75 MHz, and INOVA (100 MHz), and ³¹P NMR on a Varian Mercury 200 (81 MHz). The spectra were recorded in $CDCl_3$, DMSO- d_6 and CD_3OD solution, using TMS as the internal reference. HRMS spectra were recorded on a JEOL JMS-700. Microanalyses were registered on an Elemental VARIO EL III. Flasks, stirrings bars, and hypodermic needles used for the generation of organometallic compounds were dried for ca. 12 h at 120 °C and allowed to cool in desiccators over anhydrous calcium sulfate. Anhydrous solvents (ethers) were obtained by distillation from benzophenone ketyl.

4.1. Typical procedure for the preparation of chiral α-bromoacetamides

To a mixture of K_2CO_3 (0.77 g, 5.6 mmol) and (S)-(α -methylbenzyl)benzylamine (0.85 g, 4.0 mmol) in dichloro-

methane/water (3:2) (50 mL) at 0 °C, was added dropwise bromoacetyl bromide (0.81 g, 0.82 mL, 4.0 mmol). The reaction mixture was stirred overnight at 25 °C, quenched with water (30 mL) and extracted with dichloromethane $(3 \times 30 \text{ mL})$. The combined organic extracts were dried over anhydrous Na₂SO₄, filtered and concentrated in vacuum. The crude product was purified by flash column chromatography on silica gel (Hex/EtOAc, 9:1), obtaining (S)-2-bromoacetyl-(α -methylbenzyl)benzylamine 1 (1.29 g, 95%) as a viscous oil, $[\alpha]_{\rm D} = -115.1$ (*c* 2.8, CHCl₃). ¹H NMR (300 MHz, DMSO-*d*₆, 100 °C): δ 1.41 (d, *J* = 7.2 Hz, 3H, CH₃(CH)Ph), 4.06-4.17 (m, 3H, CH₂Br and CH₂Ph), 4.55 (AB system, J = 16.7 Hz, 1H, CH₂Ph), 5.45 (br, 1H, CH(CH₃)Ph), 7.04–7.30 (m, 10H, H_{arom}). ¹³C NMR (75 MHz, DMSO-*d*₆, 100 °C): δ 18.6 (CH₃(CH)Ph), 46.8 (CH₂Ph), 54.4 (CH(CH₃)Ph), 61.5 (CH₂Br), 127.2, 127.6, 127.8(2), 128.7, 129.0, 139.1, 141.5, 173.0 (C=O). HRMS (CI⁺, CH₄) m/z: (MH⁺) calcd for C₁₇H₁₉BrNO, 332.0650; found, 332.0656. Anal. Calcd for C₁₇H₁₈BrNO: C, 61.46; H, 5.46; N, 4.22. Found: C, 61.62; H, 5.55; N, 4.33.

4.1.1. (*S*,*S*)-2-Bromoacetyl-bis(α -methylbenzyl)amine 2. The crude product was purified by flash column chromatography on silica gel (Hex/EtOAc, 7:3), obtaining (*S*,*S*)-2 (97%) as a white solid, mp 93–94 °C, $[\alpha]_D = -127$ (*c* 3.52, CHCl₃); lit.²⁸ mp 94–95 °C, $[\alpha]_D = -125$ (*c* 3.52, CHCl₃).

4.1.2. (*S*)-Methyl 2-(2-bromoacetamide)-2-phenylacetate **3.** The crude product was purified by flash column chromatography on silica gel (Hex/EtOAc, 50:50), obtaining (*S*)-**3** (91%) as a white solid, mp 72 °C, $[\alpha]_D = +120.79$ (*c* 1.16, CHCl₃). ¹H NMR (400 MHz, CDCl₃, 25 °C): δ 3.75 (s, 3H, CO₂CH₃), 3.86 (AB system, *J* = 13.6 Hz, 1H, CH₂Br), 3.91 (AB system, *J* = 13.6 Hz, 1H, CH₂Br), 5.54 (d, *J* = 7.2 Hz, 1H, CHPh), 7.36–7.38 (m, 5H, H_{arom}), 7.48 (d, *J* = 7.2 Hz, 1H, NH). ¹³C NMR (100 MHz, CDCl₃, 25 °C): δ 28.9 (CH₂Br), 53.2 (CO₂CH₃), 57.1 (CHPh), 127.3, 128.9, 129.2, 135.8, 165.0 (CONH), 170.8 (CO₂CH₃). HRMS (CI⁺, CH₄) *m/z*: (MH⁺) calcd for C₁₁H₁₃BrNO₃, 286.0079; found, 286.0069.

4.2. Typical procedure for the preparation of the α -(dimethoxyphosphoryl)acetamides

A mixture of 1 (1.3 g, 3.9 mmol) and trimethyl phosphite (1.45 g, 1.38 mL, 11.7 mmol) without solvent was heated for 5 h at 105–110 °C. The reaction mixture was cooled and the volatile materials were evaporated in a Kugel-rohr under a reduced pressure, after which the crude product was purified by flash column chromatography (EtOAc/Hex/MeOH, 5:3:2) to give (S)-2-(dimethoxyhosphoryl-acetyl)-(α -methylbenzyl)benzylamine 4 (1.38 g, 98%) as a viscous oil, [α]_D = -80.5 (*c* 2.14, CHCl₃). ¹H NMR (72:28 rotamer ratio, asterisk denotes minor rotamer peaks, 400 MHz, CDCl₃, 25 °C): δ 1.47 (d, *J* = 7.2 Hz, 3H, CH₃CH), 1.53* (d, *J* = 6.8 Hz, 3H, CH₃CH), 2.94 (dd, *J*_{H/P} = 31 Hz, *J*_{gem} = 14.8 Hz, 1H, CH₂P), 3.22 (dd, *J*_{H/P} = 23, *J*_{gem} = 14.8 Hz, 1H, CH₂P), 3.28* (dd, *J*_{H/P} = 23, *J*_{gem} = 14.8 Hz, 1H, CH₂P), 3.78 (d, *J* = 11.2 Hz, 6H,

 $(CH_3O)_2P$), 3.83^{*} (d, J = 11.2 Hz, 3H, $(CH_3O)_2P$), 3.86^{*} (d, J = 11.2 Hz, 3H, (CH₃O)₂P), 4.01^{*} (AB system, J = 15.8 Hz, 1H, CH₂Ph), 4.31 (AB system, J = 18.4 Hz, 1H, CH₂Ph), 4.66 (AB system, J = 18.4 Hz, 1H, CH₂Ph), 4.97^{*} (AB system, J = 15.8 Hz, 1H, CH₂Ph), 5.40^{*} (q, J = 6.8 Hz, 1H, CH(CH₃)Ph), 6.15 (q, J = 7.2 Hz, 1H, CH(CH₃)Ph), 7.08–7.37 (m, 10H, H_{arom}), 7.08–7.37^{*} (m, 10H, H_{arom}). ¹³C NMR (72:28 rotamer ratio, asterisk denotes minor rotamer peaks, 100 MHz, CDCl₃, 25 °C): δ 17.3 (CH₃(CH)Ph), 19.5^{*} (CH₃(CH)Ph), 33.4^{*} (d, J =132.8 Hz, CH₂P), 33.9 (d, J = 132.8 Hz, CH₂P), 46.7* (CH₂Ph), 47.9 (CH₂Ph), 52.5 (CH(CH₃)Ph), 53.3 (d, J = 5.7 Hz, (CH₃O)₂P), 53.4^{*} (d, J = 6.9 Hz, (CH₃O)₂P), 53.5 (d, J = 6.9 Hz, (CH₃O)₂P), 53.6^{*} (d, J = 6.9 Hz, 53.5 (d, J = 0.9 Hz, (CH₃O)₂P), 55.6 (d, J = 0.9 Hz, (CH₃O)₂P), 57.3 (CH(CH₃)Ph), 125.9, 126.8 ,126.8, 127.2 ,127.4, 127.5, 127.6 ,127.9 ,128.4, 128.6 ,128.9, 137.8, 138.7 ,140.4 ,140.6, 165.7 (d, J = 6.1, (C=O)), 166.3 (d, J = 6.1, (C=O)). ³¹P NMR (121 MHz, DMSO d_6 , 100 °C): δ 24.85. HRMS (FAB⁺, CH₄) m/z: (MH⁺) calcd for C₁₉H₂₅O₄NP, 362.1521; found, 362.1510.

4.2.1. (S,S)-2-(Dimethoxyphosphorylacetyl)-bis(α -methylbenzyl)amine 5. The crude product was purified by flash column chromatography EtOAc/Hex/MeOH (5:4:1), to give (S,S)-5 (95%) as viscous oil, $[\alpha]_D = -112.3$ (c 2.02, CHCl₃). ¹H NMR (300 MHz, DMSO-d₆, 120 °C): δ 1.69 (d, J = 6.9 Hz, 3H, CH₃(CH)Ph), 3.08 (dd, $J_{H/P} = 21.3$, $J_{gem} = 9.0$ Hz, 1H, CH₂P), 3.11 (dd, $J_{H/P} = 21.3$, $J_{gem} =$ 9.0 Hz, 1H, CH₂P), 3.64 (d, J = 10.9 Hz, 3H, (CH₃O)₂P), 3.65 (d, J = 10.9 Hz, 3H, (CH₃O)₂P), 5.05 (br, 1H, CH(CH₃)Ph), 7.09–7.14 (m, 10H, H_{arom}). ¹³C NMR (75 MHz, DMSO-d₆, 120 °C): δ 19.4 (CH₃(CH)Ph), 34.9 (d, J = 133.2 Hz, CH₂P), 53.2 (d, J = 6.2 Hz, (CH₃O)₂P), 53.3 (d, J = 6.2 Hz, (CH₃O)₂P), 54.9 (CH(CH₃)Ph), 127.3, 128.1, 128.3, 141.7, 165.0 (d, J = 5.1, (C=O)). ³¹P NMR (121 MHz, DMSO-d₆, 100 °C): δ 25.86. HRMS $(CI^+, CH_4) m/z$: (MH^+) calcd for $C_{20}H_{27}NO_4P$, 376.1678; found, 376.1668.

4.2.2. (S)-Methyl 2-[2-(dimethoxyphosphoryl)acetamide]2phenylacetate 6. The crude product was purified by flash column chromatography (EtOAc/Hex/MeOH, 5:3:2) to give the chiral phosphonoacetamide (S)-6 (92%) as a white solid, mp 66–67 °C, $[\alpha]_D = +122.75$ (*c* 1.06, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 2.90 (dd, $J_{H/P} = 19.0$ Hz, $\begin{aligned} J_{gem} &= 15.6 \text{ Hz}, \quad 1\text{H}, \quad \text{CH}_2\text{P}), \quad 2.96 \quad (\text{dd}, \quad J_{\text{H/P}} = 19.0, \\ J_{gem} &= 15.6 \text{ Hz}, \quad 1\text{H}, \quad \text{CH}_2\text{P}), \quad 3.71 \quad (\text{d}, \quad J = 10.8 \text{ Hz}, \quad 3\text{H}, \end{aligned}$ $(CH_3O)_2P$, 3.72 (s, 3H, CO₂CH₃), 3.82 (d, J = 10.8 Hz, 3H, $(CH_3O)_2P$), 5.55 (d, J = 7.0 Hz, 1H, CHPh), 7.30– 7.41 (m, 5H, H_{arom}), 7.74 (d, J = 7.0 Hz, 1H, NH). ¹³C NMR (100 MHz, CDCl₃): δ 34.4 (d, $J_{C/P} = 131.3$ Hz, CH₂P), 53.0 (CO₂CH₃), 53.3 (d, J = 6.1 Hz, (CH₃O)₂P), 53.5 (d, J = 6.1 Hz, (CH₃O)₂P), 57.1 (CHPh), 127.4, 128.7, 129.1, 136.0, 163.3 (d, J = 3.8 Hz, CONH), 170.9 (CO_2CH_3) . ³¹P NMR (81 MHz, CDCl₃): δ 25.86. HRMS $(CI^+, CH_4) m/z$: (MH^+) calcd for $C_{13}H_{19}NO_6P$, 316.0950; found, 316.0943.

4.2.3. (S)-2-(2-Dimethoxyphosphoryl)acetamide-2-phenylethanol 7. To a solution of 6 (0.4 g, 1.27 mmol) in dry THF (20 mL) at 0 °C, sodium borohydride 290 mg, (7.62 mmol) was added. The resulting solution was heated

for 20 min at reflux, and then methanol (4 mL) was added dropwise and refluxed for an additional period of 60 min. After this time, the reaction mixture was cooled and an HCl/i-PrOH solution was added until pH 3-4. The solid was filtered and purified by column chromatography (EtOAc/Hex/i-PrOH, 5:3:3), to obtain the chiral phosphonoacetamide (S)-7 (0.32 g, 89%) as a viscous oil, $[\alpha]_{D} = +34.5$ (c 2.98, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 2.93 (dd, $J_{H/P} = 16.8$, $J_{gem} = 14.8$ Hz, 1H, CH₂P), 2.87 (dd, $J_{H/P} = 16.8$, $J_{gem} = 14.8$ Hz, 1H, CH₂P), 3.65 (d, J = 11.4 Hz, 3H, (CH₃O)₂P), 3.73 (ddd, $J_{gem} =$ 11.6, J = 5.0, J = 1.2 Hz, 1H, CH₂OH), 3.74 (d, J = 11.4 Hz, 3H, (CH₃O)₂P), 3.80 (ddd, $J_{gem} = 11.6$, J = 7.6, J = 1.2 Hz, 1H, CH₂OH), 5.04 (ddd, J = 7.6,J = 5.0 Hz, 1H, CHPh), 7.18–7.30 (m, 5H, H_{aron}), 7.65 (d, J = 7.6 Hz, 1H, NH). ¹³C NMR (100 MHz, CDCl₃): δ 34.8 (d, J = 130.5 Hz, CH₂P), 53.3 (d, J = 6.8 Hz, (CH₃O)₂P), 53.7 (d, J = 6.8 Hz, (CH₃O)₂P), 66.0 (CH₂OH), 126.8, 127.7, 128.7, 139.0, 164.1 (d, J = 4.6 Hz, CONH). ³¹P NMR (81 MHz, CDCl₃): δ 26.77; HRMS (CI⁺, CH₄) m/z: (MH⁺) calcd for C₁₂H₁₉NO₅P, 288.0995; found, 288.0979.

(S)-2-(2-Diethoxyphosphoryl)acetamide-2-phenvl-4.2.4. ethanol 8. A solution of dicyclohexylcarbodiimide (DCC) (1.51 g, 7.3 mmol), 4-(N,N-dimethylamino)pyridine (DMAP) (80 mg, 0.66 mmol) in 70 mL of dry dichloromethane was slowly added under an inert atmosphere at 0 °C to a solution of L-phenylglycinol (1.0 g, 7.3 mmol) 2-(diethoxyphosphoryl)acetic acid 10 (1.57 g, and 8.03 mmol) in 70 mL of dry dichloromethane. The reaction mixture was stirred for 12 h at room temperature, filtered and concentrated in vacuum. The crude product was purified by column chromatography (EtOAc/Hex/i-PrOH, 5:3:3) obtaining phosphonoacetamide (S)-8 (2.14 g, 93%yield) as a viscous oil, $[\alpha]_D = +41.8$ (c 2.88, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 1.25 (t, J = 6.8 Hz, 3H, $(CH_3CH_2O)_2P$, 1.31 (t, J = 6.8 Hz, 3H, $(CH_3CH_2O)_2P$), 2.91 (dd, $J_{H/P} = 18.4$, $J_{gem} = 14.4$ Hz, 1H, CH₂P), 2.96 (dd, $J_{H/P} = 18.4$, $J_{gem} = 14.4$ Hz, 1H, CH₂P), 3.77 (ddd, $J_{gem} = 10.8$, J = 6.8, J = 1.2 Hz, 1H, CH₂OH), 3.82 (ddd, $J_{gem} = 10.8, J = 4.4, J = 1.2$ Hz, 1H, CH₂OH), 4.05 (dq, $J_{H/P}^{s.m} = 14.0, J = 6.8$ Hz, 2H, CH₂OP), 4.14 (dq, $J_{H/P} =$ 14.0, J = 6.8 Hz, 2H, CH₂O–P), 5.05–5.09 (m, 1H, CHPh), 7.23–7.35 (m, 5H, H_{arom}), 7.72 (br, 1H, NH). ¹³C NMR (100 MHz, CDCl₃): δ 16.5 ((CH₃CH₂O)₂P), 16.6 $((CH_3CH_2O)_2P)$, 35.0 (d, J = 129.0 Hz, $CH_2P)$, 56.3 (CH_2OH) , 63.0 (d, J = 6.8 Hz), (CH_2OP) , 63.4 (d, J = 6.8 Hz, CH₂OP), 65.9 (CH₂Ph), 126.8, 127.6, 128.6, 139.1, 164.4 (CONH).³¹P NMR (81 MHz, CDCl₃): δ 24.07. HMRS (CI⁺, CH₄) m/z: (MH⁺) calcd for C₁₄H₂₃NO₅P, 316.1314; found, 316.1301.

4.3. General procedure for the HWE reaction

A solution of phosphonoacetamides 4-8 (1.0 equiv) in dry THF (30 mL) was treated under a nitrogen atmosphere with 8-diazabicyclo[5.4.0]undec-7-ene (DBU) (3.0 equiv) and lithium chloride (3.0 equiv) at room temperature. After stirring for 5 min, the corresponding aldehyde (1.0 equiv) was added and the reaction mixture stirred for 4 h at room temperature. The reaction was quenched by the addition of

a saturated NH₄Cl solution, and extracted with ethyl acetate (3 × 40 mL). The combined extracts were washed with water (30 mL) followed by brine (30 mL), dried over Na₂SO₄ and concentrated in vacuum. The *E*:*Z* ratio was determined by ¹H NMR on the crude product, and then the (*S*)-(*E*)- α , β -unsaturated amides were purified by flash chromatography.

4.3.1. (*S*)-(*E*)-*N*-(Cinnamoyl)-(α -methylbenzyl)benzylamine **11a.** (350 mg, 93%), as a viscous oil, $[\alpha]_{D} = -182.8$ (*c* 4.80, CHCl₃). ¹H NMR (300 MHz, DMSO-*d*₆, 100 °C): δ 1.49 (d, *J* = 6.6 Hz, 3H, CH₃(CH)Ph), 4.36 (AB system, *J* = 17.0 Hz, 1H, CH₂Ph), 4.78 (AB system, *J* = 17.0 Hz, 1H, CH₂Ph), 5.81 (q, *J* = 6.6 Hz, 1H, CH(CH₃)Ph), 7.02 (d, *J*_{trans} = 15.4 Hz, 1H, CHC=O), 7.18–7.50 (m, 15H_{aron}), 7.56 (d, *J*_{trans} = 15.4 Hz, 1H, CHC=O), 7.18–7.50 (m, 15H_{aron}), 7.56 (d, 100 °C): δ 18.8 (CH₃(CH)Ph), 47.6 (CH₂Ph), 54.4 (CH(CH₃)Ph), 120.6, 127.2, 127.3, 127.7, 127.7, 128.3, 128.8, 128.9, 129.3 129.9, 135.9, 140.0, 141.9, 142.2, 167.2 (C=O). HRMS (CI⁺, CH₄) *m*/*z*: (MH⁺) calcd for C₂₄H₂₄NO, 342.1858; found, 342.1865.

4.3.2. (*S*)-(*E*)-*N*-(*p*-Chlorocinnamoyl)-(α -methylbenzyl)benzylamine **11b.** (390 mg, 94%), as a viscous oil, $[\alpha]_{D} = -170.7$ (*c* 3.03 CHCl₃). ¹H NMR (300 MHz, DMSO-*d*₆, 100 °C): δ 1.49 (d, *J* = 6.9 Hz, 3H, CH₃(CH)Ph), 4.35 (AB system, *J* = 16.9 Hz, 1H, CH₂Ph), 4.78 (AB system, *J* = 16.9 Hz, 1H, CH₂Ph), 5.81 (q, *J* = 6.9 Hz, 1H, CH(CH₃)Ph), 7.10 (d, *J*_{trans} = 15.1 Hz, 1H, CHC=O), 7.14–7.56 (m, 14H_{arom}) 7.54 (d, *J*_{trans} = 15.1 Hz, 1H, CH(*p*-ClC₆H₄)). ¹³C NMR (75 MHz, DMSO-*d*₆, 100 °C): δ 18.6, (CH₃(CH)Ph), 47.6 (CH₂Ph), 54.6 (CH(CH₃)Ph), 121.2, 127.2, 127.3, 127.6, 127.8, 128.8, 129.0, 129.3, 129.8, 134.6, 134.8, 139.6, 140.7, 141.8, 167.3 (C=O). HRMS (CI⁺, CH₄) *m/z*: (MH⁺) calcd for C₂₄H₂₃ClNO, 376.1468; found, 376.1466.

4.3.3. (*S*)-(*E*)-*N*-(*p*-Methoxycinnamoyl)-(α-methylbenzyl)benzylamine 11c. (400 mg, 97%), as a viscous oil, $[α]_D = -185.3$ (*c* 2.42, CHCl₃). ¹H NMR (400 MHz, DMSO-*d*₆, 120 °C): δ 1.51 (d, *J* = 6.6 Hz, 3H, CH₃(CH)Ph), 3.78 (s, 3H, CH₃O), 4.37 (AB system, *J* = 16.8 Hz, CH₂Ph), 4.78 (AB system, *J* = 16.8 Hz, CH₂Ph), 5.82 (q, *J* = 6.6 Hz, 1H, CH(CH₃)Ph), 6.91 (AA'BB' system, *J* = 8.8 Hz, 2H, H_{arom}), 7.20 (d, *J*_{trans} = 15.2 Hz, 1H, CHC=O), 7.17-7.46 (m, 12H, H_{arom}), 7.54 (d, *J*_{trans} = 15.2 Hz 1H, CHAr). ¹³C NMR (100 MHz, DMSO-*d*₆, 120 °C): δ 17.4 (CH₃(CH)Ph), 46.3 (CH₂Ph), 54.8 (CH(CH₃)Ph), 54.8 (CH₃O), 113.8, 116.8, 125.8, 126.0, 126.3, 126.4, 127.4, 127.6, 128.6, 138.9, 140.5, 140.6, 141.0, 160.0, 166.1 (C=O). HRMS (Cl⁺, CH₄) *m/z*: (MH⁺) calcd for C₂₅H₂₆NO₂, 372.1964; found, 372.1956.

4.3.4. (*S*)-(*E*)-*N*-(3-*tert*-Butylacryloyl)-(α -methylbenzyl)benzylamine 11d. (330 mg, 92%), as a white solid, mp 63–64 °C, [α]_D = -156.3 (*c* 2.63, CHCl₃). ¹H NMR (300 MHz, DMSO-*d*₆, 100 °C): δ 0.92 (s, 9H, (CH₃)₃C), 1.46 (d, *J* = 6.9 Hz, 3H, CH₃(CH)Ph), 4.31 (AB system, *J* = 16.8 Hz, 1H, CH₂Ph), 4.64 (AB system, *J* = 16.8 Hz, 1H, CH₂Ph), 5.69 (br, 1H, CH(CH₃)Ph), 6.08 (d, *J*_{trans}= 15.3 Hz, 1H, CHC=O), 6.68 (d, *J*_{trans} = 15.3 Hz, 1H, CHC

CH(CH₃)₃), 7.10–7.33 (m, 10H, H_{arom}). ¹³C NMR (75 MHz, DMSO- d_6 , 100 °C): δ 18.6(CH₃(CH)Ph), 29.4 ((CH₃)₃C), 33.8 (C(CH₃)₃), 47.8 (CH₂Ph), 54.2 (CH(CH₃)Ph), 118.6, 127.2, 127.3, 127.6, 127.7, 128.7, 128.9, 139.9, 142.2, 155.3, 167.8 (C=O). HRMS (Cl⁺, CH₄) *m*/*z*: (MH⁺) calcd for C₂₂H₂₈NO, 322.2171; found, 322.2166.

4.3.5. (S)-(E)-N-(3-iso-Propylacryloyl)- $(\alpha$ -methylbenzyl)benzylamine 11e. (320 mg, 94%), as a viscous oil. ¹H NMR (300 MHz, DMSO- d_6 , 100 °C): δ 0.91 (d, J = 6.8 Hz, 3H, (CH₃)₂CH), 0.92 (d, J = 6.8 Hz, 3H, $(CH_3)_2CH$, 1.45 (d, J = 7.2 Hz, 3H, $CH_3(CH)Ph$), 2.31– 2.38 (m, 1H, CH(CH₃)₂), 4.28 (AB system, J = 16.8 Hz, 1H, CH₂Ph), 4.65 (AB system, J = 16.8 Hz, 1H, CH₂Ph), 5.69 (br, 1H, CH(CH₃)Ph), 6.19 (d, $J_{trans} = 15.1$ Hz, 1H, CHC=O), 6.70 (dd, $J_{trans} = 15.1$ Hz, J = 6.8 Hz, 1H, CH*i*-Pr), 7.15–7.34 (m, 10H, H_{arom}). ¹³C NMR (100 MHz, DMSO-d₆, 100 °C): δ 18.7 (CH₃(CH)Ph), 22.1 ((CH₃)₂CH), 30.9, (CH(CH₃)₂), 47.6 (CH₂Ph), 54.2 (CH(CH₃)Ph), 120.3, 127.1, 127.3, 127.6, 128.7, 128.9, 128.9, 140.0, 142.3, 151.8, 167.4 (C=O). HRMS (Cl⁺, CH₄) m/z: (MH⁺) calcd for C₂₁H₂₆NO, 308.2014; found, 308.2009.

4.3.6. (S)-(E)-N-(3-iso-Butylacryloyl)-(α-methylbenzyl)benzylamine 11f. (340 mg, 95%), as a viscous oil, $[\alpha]_{\rm D} = -142.8$ (c 2.0, CHCl₃). ¹H NMR (300 MHz, DMSO- d_6 , 100 °C): δ 0.81 (d, J = 6.6 Hz, 3H, (CH₃)₂CH), 0.82 (d, J = 6.6 Hz, 3H, (CH₃)₂CH), 1.44 (d, J = 7.2 Hz, 3H, CH₃(CH)Ph), 1.64 (n, J = 6.6 Hz, 1H, CH(CH₃)₂), 2.01 (t, J = 6.6 Hz, 2H, CH₂CH(CH₃)₂), 4.26 (AB system, J = 17.1 Hz, 1H, CH₂Ph), 4.65 (AB system, J = 17.1 Hz, 1H, CH₂Ph), 5.69 (br, 1H, CH(CH₃)Ph), 6.27 (d, $J_{trans} = 14.7$ Hz, 1H, CHC=O), 6.72 (dt, $J_{trans} = 15.0$ Hz, J = 7.2 Hz, 1H, CHCH₂*i*-Pr), 7.10–7.34 (m, 10H, H_{arom}). ¹³C NMR (100 MHz, DMSO-*d*₆, 100 °C): δ 18.7 (CH₃(CH)Ph), 22.7 ((CH₃)₂CH), 22.8 ((CH₃)₂CH), 28.1 (CH₂CH(CH₃)₂), 41.6 (CH(CH₃)₂), 47.4 (CH₂Ph), 54.1 (CH(CH₃)Ph), 123.9, 127.1, 127.2, 127.6, 127.7, 128.7, 128.9, 140.0, 142.2, 144.3, 167.2 (C=O). HRMS (Cl⁺, CH₄) m/z: (MH⁺) calcd for C₂₂H₂₈NO, 322.2171; found, 322.2163.

4.3.7. (S,S)-(E)-N-(Cinnamoyl)-bis $(\alpha$ -methylbenzyl)amine 12a. (352 mg, 93%), as a white solid, mp 151-152 °C, $[\alpha]_{D} = +6.8$ (c 3.22, CHCl₃). ¹H NMR (400 MHz, $DMSO-d_6$, 110 °C): δ 1.75 (d, J = 7.2 Hz, 6H. CH₃(CH)Ph), 5.38 (br, 2H, CH(CH₃)Ph), 6.71 (d, $J_{trans} = 15.8 \text{ Hz}, 1\text{H}, \text{CHC}=0$, 7.16–7.36 (m, 15H, H_{arom}), 7.40 (d, $J_{trans} = 15.8$ Hz, 1H, CHPh). ¹³C NMR (100 MHz. DMSO-*d*₆, 110 °C): δ 18.9 $(CH_3(CH)Ph),$ 52.5 (CH(CH₃)Ph), 121.5, 126.5, 127.0, 127.1, 127.6, 128.4, 128.9, 135.1, 139.6, 141.4, 165.4 C=O. HRMS (FAB⁺) CH₄) m/z: (MH⁺) calcd for C₂₅H₂₆NO, 356.2014; found, 356.2010. Anal. Calcd for C25H25NO: C, 84.47; H, 7.09; N, 3.94. Found: C, 84.05; H, 7.14; N, 4.14.

4.3.8. (*S*,*S*)-(*E*)-*N*-(*p*-Chlorocinnamoyl)-bis(α -methylbenzyl)amine 12b. (390 mg, 94%), as a white solid, mp 123– 124 °C, [α]_D = +48.3 (*c* 3.66, CHCl₃). ¹H NMR (300 MHz, DMSO-*d*₆, 100 °C): δ 1.69 (d, *J* = 6.9 Hz, 6H, CH₃(CH)Ph), 5.31 (br, 2H, CH(CH₃)Ph), 6.61 (d, $J_{trans} = 15.4$ Hz, 1H, CHC=O), 7.08–7.30 (m, 14H, H_{arom}), 7.28 (d, $J_{trans} = 15.4$ Hz, 1H, CH(*p*-ClC₆H₄)). ¹³C NMR (75 MHz, DMSO-*d*₆, 100 °C): δ 19.7 (CH₃(CH)Ph), 53.5 (CH(CH₃)Ph), 123.1, 127.6, 127.8, 128.6, 129.3, 129.6, 134.6, 134.7, 139.2, 142.0, 166.4 C=O. HRMS (FAB⁺, CH₄) *m*/*z*: (MH⁺) calcd for C₂₅H₂₅ClNO, 390.1625; found, 390.1638. Anal. Calcd for C₂₅H₂₄ClNO: C, 77.01; H, 6.20; N, 3.59. Found: C, 76.98; H, 6.29; N, 3.69.

4.3.9. (*S*,*S*)-(*E*)-*N*-(*p*-Methoxycinnamoyl)-bis(α -methylbenzyl)amine 12c. (374 mg, 91%), as a white solid, mp 102– 103 °C, [α]_D = +56.4 (*c* 3.77, CHCl₃). ¹H NMR (400 MHz, DMSO-*d*₆, 120 °C): δ 1.75 (d, *J* = 7.2 Hz, 6H, CH₃(CH)Ph), 3.77 (s, 3H, CH₃O), 5.37 (q, *J* = 7.2 Hz, 2H, CH(CH₃)Ph), 6.57 (d, *J*_{trans} = 15.2 Hz, 1H, CHC=O), 6.88 (AA'BB' system, *J* = 8.4 Hz, 2H, H_{arom}), 7.14–7.21 (m, 10H, H_{arom}), 6.30 (AA'BB' system, *J* = 8.4 Hz, 2H, H_{arom}), 7.38 (d, *J*_{trans} = 15.2 Hz, 1H, CH(*p*-MeOC₆H₄)). ¹³C NMR (100 MHz, DMSO-*d*₆, 120 °C): δ 19.1 (CH₃(CH)Ph), 52.7 (CH(CH₃)Ph), 55.4 (CH₃O), 114.4, 119.4, 126.6, 127.1, 127.8, 128.0, 128.9, 139.6, 141.7, 160.4, 165.9 C=O. HRMS (FAB⁺, CH₄) *m/z*: (MH⁺) calcd for C₂₆H₂₈NO₂, 386.2120; found, 386.2137. Anal. Calcd for C₂₆H₂₇NO₂: C, 81.01; H, 7.06; N, 3.63. Found: C, 81.08; H, 7.18; N, 3.75.

4.3.10. (*S*,*S*)-(*E*)-*N*-(3-*tert*-Butylacryloyl)-bis(α -methylbenzyl)amine 12d. (330 mg, 92%), as a white solid, mp 114 °C, [α]_D = -170.9 (*c* 3.64, CHCl₃). ¹H NMR (300 MHz, DMSO-*d*₆, 100 °C): δ 0.79 (s, ((CH₃)₃C)), 1.65 (d, *J* = 7.2 Hz, 6H, CH₃(CH)Ph), 5.31 (br, 2H, CH(CH₃)Ph), 5.67 (d, *J*_{trans} = 15.4 Hz, 1H, CH=CO), 6.49 (d, *J*_{trans} = 15.4 Hz, 1H, CHC(CH₃)₃), 7.08–7.19 (m, 10H, H_{arom}). ¹³C NMR (75 MHz, DMSO-*d*₆, 100 °C): δ 19.7 (CH₃(CH)Ph), 29.3 ((CH₃)₃C), 33.6 (C(CH₃)₃), 52.8 (CH(CH₃)Ph), 120.4, 127.5, 127.7, 128.6, 142.2, 154.0, 167.2 C=O. HRMS (FAB⁺, CH₄) *m/z*: (MH⁺) calcd for C₂₃H₃₀ON, 336.2327; found, 336.2332. Anal. Calcd for C₂₃H₂₉ON: C, 82.34; H, 8.71; N, 4.18. Found: C, 82.54; H, 8.81; N, 4.22.

4.3.11. (*S*,*S*)-(*E*)-*N*-(3-*iso*-Propylacryloyl)-bis(α -methylbenzyl)amine 12e. (340 mg, 99%), as a viscous oil. ¹H NMR (300 MHz, DMSO-*d*₆, 100 °C): δ 0.82 (d, *J* = 6.7 Hz, 3H, (CH₃)₂CH), 0.84 (d, *J* = 6.7 Hz, 3H, (CH₃)₂CH), 1.68 (d, *J* = 7.2 Hz, 6H, CH₃(CH)Ph), 2.20–2.27 (m, 1H, CH(CH₃)₂), 5.30 (q, *J* = 7.2 Hz, 2H, CH(CH₃)Ph), 5.87 (dd, *J*_{trans} = 15, *J*_{allylic} = 1.2 Hz, 1H, CHC=O), 6.55 (dd, *J*_{trans} = 15, *J* = 6.7 Hz, 1H, CH*i*-Pr), 7.10–7.20 (m, 10H, H_{arom}). ¹³C NMR (75 MHz, DMSO-*d*₆, 100 °C): δ 19.7 (CH₃(CH)Ph), 21.9 ((CH₃)₂CH), 22.0 ((CH₃)₂CH), 30.7 (CH(CH₃)₂), 53.0 (CH(CH₃)Ph), 122.2, 127.3, 127.8, 128.5, 142.4, 150.2, 166.7 C=O. HRMS (CI⁺, CH₄) *m/z*: (MH⁺) calcd for C₂₂H₂₈ON, 322.2171; found, 322.2169.

4.3.12. (*S*,*S*)-(*E*)-*N*-(3-*iso*-Butylacryloyl)-bis(α -methylbenzyl)amine 12f. (360 mg, 99%), as a white solid, mp 64– 65 °C, [α]_D = -81.7 (*c* 1.66, CHCl₃). ¹H NMR (300 MHz, DMSO-*d*₆, 100 °C): δ 0.78 (d, *J* = 6.7 Hz, 3H, (CH₃)₂CH), 0.79 (d, *J* = 6.7 Hz, 3H, (CH₃)₂CH), 1.59 (m, 1H, CH(CH₃)₂), 1.69 (d, *J* = 6.9 Hz, 6H, CH₃(CH)Ph), 1.90 (m, 2H, CH₂*i*-Pr), 5.27 (q, J = 6.9 Hz, 2H, CH(CH₃)Ph), 6.00 (dt, $J_{trans} = 15$, $J_{allylic} = 1.5$ Hz, 1H, CHC=O), 6.57 (dt, $J_{trans} = 15$, J = 7.2 Hz, 1H, CH*i*-Bu), 7.09–7.20 (m, 10H, H_{arom}). ¹³C NMR (75 MHz, DMSO-*d*₆, 100 °C): δ 19.8 (CH₃(CH)Ph), 22.7 ((CH₃)₂CH), 22.8 ((CH₃)₂CH), 28.0 (CH₂*i*-Pr), 41.6 (CH(CH₃)₂), 53.2 ((CH(CH₃)Ph)), 125.9, 127.3, 127.8, 128.5, 142.3, 142.7, 166.4 C=O. HRMS (CI⁺, CH₄) *m/z*: (MH⁺) calcd for C₂₃H₃₀ON, 336.2327; found, 336.2342.

4.3.13. (*S*)-(*E*)-Methyl *N*-(cinnamoyl)-2-amino-2-phenylacetate 13a. (150 mg, 82%), as a white solid, mp 158 °C, $[\alpha]_D = +39.9$ (*c* 1.26, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 3.75 (s, 3H, CO₂CH₃), 5.74 (d, *J* = 6.8 Hz, 1H, CHPh), 6.49 (d, *J*_{trans} = 15.4 Hz, 1H, CHCO), 6.78 (d, *J* = 6.8 Hz, 1H, NH), 7.33–7.49 (m, 10H, H_{arom}), 7.63 (d, *J*_{trans} = 15.4 Hz, 1H, CHPh). ¹³C NMR (100 MHz, CDCl₃): δ 53.1 (CO₂CH₃), 56.7 (CHPh), 119.9, 127.5, 128.1, 128.8, 128.9, 129.2, 130.1, 134.8, 136.7, 142.3, 165.3 (CONH), 171.7 (CO₂Me). HRMS (CI⁺, CH₄) *m/z*: (MH⁺) calcd for C₁₈H₁₈NO₃, 296.1281; found, 296.1287. Anal. Calcd for C₁₈H₁₇NO₃: C, 73.20; H, 5.80; N, 4.74. Found: C, 73.17; H, 5.86; N, 4.81.

4.3.14. (*S*)-(*E*)-Methyl *N*-(*p*-chlorocinnamoyl)-2-amino-2phenylacetate 13b. (289 mg, 93%), as a white solid, mp 183 °C, $[\alpha]_D = +69.8$ (*c* 1.1 CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 3.74 (s, 3H, CO₂CH₃), 5.71 (d, *J* = 7.0 Hz, 1H, CHPh), 6.46 (d, *J*_{trans} = 15.8 Hz, 1H, CHCO), 6.87 (d, *J* = 7.0 Hz, 1H, NH), 7.29–7.40 (m, 9H, H_{arom}), 7.56 (d, *J*_{trans} = 15.8 Hz, 1H, CHAr). ¹³C NMR (100 MHz, CDCl₃): δ 53.1 (CO₂CH₃), 56.7 (CHPh), 120.5, 127.5, 128.8, 129.1, 129.2, 129.2, 133.2, 135.8, 136.5, 140.9, 165.0 (CONH), 171.7 (*C*O₂CH₃). HRMS (CI⁺, CH₄) *m/z*: (MH⁺) calcd for C₁₈H₁₇CINO₃, 330.0891; found, 330.0892. Anal. Calcd for C₁₈H₁₆CINO₃: C, 65.56; H, 4.89; N, 4.25. Found: C, 65.26; H, 4.94; N, 4.23.

4.3.15. (*S*)-(*E*)-Methyl *N*-(*p*-methoxycinnamoyl)-2-amino-2phenylacetate 13c. (271 mg, 86%), as a white solid, mp 146–147 °C, $[\alpha]_D = -2.2$ (*c* 1.1, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 3.74 (s, 3H, CO₂CH₃), 3.81 (s, 3H, CH₃OAr), 5.73 (d, *J* = 7.0 Hz, 1H, CHPh), 6.36 (d, *J*_{trans} = 15.6 Hz, 1H, CHCO), 6.74 (d, *J* = 7.0 Hz, 1H, NH), 6.86 (AA'BB' system, *J* = 8.8 Hz, 2H, H_{arom}), 7.33– 7.44 (m, 7H, H_{arom}), 7.59 (d, *J*_{trans} = 15.6 Hz, 1H, CHAr). ¹³C NMR (100 MHz, CDCl₃): δ 53.0 (CO₂CH₃), 56.7 (CHPh), 114.4, 117.5, 127.4, 127.5, 128.7, 129.1, 129.6, 136.8, 141.8, 161.2, 165.7 (CONH), 171.8 (CO₂CH₃). HRMS (CI⁺, CH₄) *m/z*: (MH⁺) calcd for C₁₉H₂₀NO₄, 326.1387; found, 326.1379. Anal. Calcd for C₁₉H₁₉NO₄: C, 70.14; H, 5.89; N, 4.31. Found: C, 69.98; H, 5.90; N, 4.38.

4.3.16. (*S*)-(*E*)-Methyl *N*-(3-*tert*-butylacryloyl)-2-amino-2phenylacetate 13d. (184 mg, 70%), as a white solid, mp 133–134 °C, $[\alpha]_D = -1.1$ (*c* 2.8, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 1.05 (s, 9H, (CH₃)₃C), 3.74 (s, 3H, CO₂CH₃), 5.68 (d, *J* = 7.0 Hz, 1H, CHPh), 5.75 (d, *J*_{trans} = 15.4 Hz, 1H, CHCO), 6.56 (d, *J* = 7.0 Hz, 1H, NH), 6.88 (d, *J*_{trans} = 15.4 Hz, 1H, CH*t*-Bu), 7.31–7.40, (m, 5H, H_{arom}). ¹³C NMR (100 MHz, CDCl₃): δ 28.9 ((CH₃)₃CH), 33.7 (CH(CH₃)₃), 53.0 (CO₂CH₃), 56.5 (CHPh), 118.1, 127.4, 128.6, 129.0, 136.6, 156.0, 165.6 (CONH), 171.6 (CO₂CH₃). HRMS (CI⁺, CH₄) m/z: calcd for (MH⁺) C₁₆H₂₂NO₃, 276.1600; found, 276.1606. Anal. Calcd for C₁₆H₂₁NO₃: C, 69.79; H, 7.69; N, 5.09. Found: C, 69.79; H, 7.76; N, 5.19.

4.3.17. (*S*)-(*E*)-13e and (*S*)-(*Z*)-Methyl *N*-(3-*iso*-propylacry-loyl)-2-amino-2-phenylacetate 13e.

4.3.17.1. (*S*)-(*E*)-Isomer 13e (more polar). (143 mg, 56%), as a white solid, mp 98–100 °C, $[\alpha]_{D} = +0.7$ (*c* 2.37, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 1.03 (d, J = 7.0 Hz, 6H, (CH₃)₂CH), 2.42 (m, 1H, CH(CH₃)₂), 3.73 (s, 3H, CO₂CH₃), 5.67 (d, J = 7.2 Hz, 1H, CHPh), 5.80 (dd, $J_{trans} = 15.4$, $J_{allylic} = 1.5$ Hz, 1H, CHCO), 6.59 (d, J = 7.2 Hz, 1H, NH), 6.85 (dd, $J_{trans} = 15.4$, $J_{allylic} = 1.5$ Hz, 1H, CHCO), 6.59 (d, J = 7.0 Hz, 1H, CHi-Pr), 7.31–7.39 (m, 5H, H_{arom}). ¹³C NMR (100 MHz, CDCl₃): δ 21.6 ((CH₃)₂CH), 21.6 ((CH₃)₂CH), 30.9 (CH(CH₃)₂), 53.0 (CHPh), 56.5 (CO₂CH₃), 120.1, 127.4, 128.5, 129.0, 136.6, 152.3, 165.4 (CONH), 171.5 (CO₂CH₃). HRMS (CI⁺, CH₄) *m/z*: (MH⁺) calcd for C₁₅H₂₁NO₃, 262.1438; found, 262.1461. Anal. Calcd for C₁₅H₂₀NO₃: C, 68.93; H, 7.33; N, 5.36. Found: C, 68.80; H, 7.27; N, 5.43.

4.3.17.2. (S)-(Z)-Isomer 13e (less polar). (88 mg, 34%), as a white solid, mp 125–126 °C, $[\alpha]_D = +7.2$ (c 1.36, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 0.97 (d, J = 6.8 Hz, 3H, (CH₃)₂CH), 0.99 (d, J = 6.8 Hz, 3H, (CH₃)₂CH), 3.57–3.69 (m, 1H, CH(CH₃)₂), 3.73 (s, 3H, CO_2CH_3), 5.61 (d, J = 5.6 Hz, 1H, CHPh), 5.63 (dd, $J_{cis} = 10.4$, $J_{allylic} = 0.8$ Hz, 1H, CHCO), 5.85 (dd, $J_{cis} = 10.4, J = 11.6$ Hz, 1H, CH*i*-Pr), 6.50 (d, J = 5.6 Hz, 1H, NH), 7.31–7.39 (m, 5H, H_{arom}). ¹³C NMR (100 MHz, CDCl₃): δ 22.7 ((CH₃)₂CH), 22.8 ((CH₃)₂CH), 27.6 (CH(CH₃)₂), 53.0 (CHPh), 56.4 (CO₂CH₃), 119.0, 127.4, 128.6, 129.0, 136.7, 154.1, 165.4 (CONH), 171.5 (CO₂CH₃). HRMS (CI⁺, CH₄) m/z: (MH⁺) calcd for C₁₅H₂₁NO₃, 262.1438; found, 262.1453. Anal. Calcd for C₁₅H₂₀NO₃: C, 68.93; H, 7.33; N, 5.36. Found: C, 68.78; H, 7.31; N, 5.41.

4.3.18. (S)-(E)-13f and (S)-(Z)-Methyl N-(3-iso-butylacry-loyl)-2-amino-2-phenylacetate 13f.

4.3.18.1. (S)-(E)-Isomer 13f (more polar). (203 mg, 75%), as a white solid, mp 100 °C, $[\alpha]_D = +86.15$ (c 0.78, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 0.90 (d, J = 7.0 Hz, 3H, (CH₃)₂CH), 0.91 (d, J = 7.0 Hz, 3H, $(CH_3)_2CH$, 1.73 (m, 1H, CH $(CH_3)_2$), 2.05 (dt, J = 7.0, 1.0 Hz, 2H, CH₂*i*-Pr), 3.73 (s, 3H, CO₂CH₃), 5.66 (d, J = 7.2 Hz, 1H, CHPh), 5.84 (dt, $J_{trans} = 15.1$, $J_{allylic} =$ 1.2 Hz, 1H, CHC=O), 6.55 (d, J = 7.2 Hz, 1H, NH), 6.85 $(dt, J_{trans} = 15.1, J = 7.0 \text{ Hz}, 1\text{H}, \text{CH}i\text{-Bu}), 7.30-7.39 \text{ (m},$ 5H, H_{arom}). ¹³C NMR (100 MHz, CDCl₃): δ 22.6 ((CH₃)₂CH), 22.6 ((CH₃)₂CH), 28.0 (CH(CH₃)₂), 41.6 (CH₂*i*-Pr), 53.0 (CHPh), 56.5 (CO₂CH₃), 123.8, 127.4, 128.6, 129.0, 136.6, 145.1, 165.1 (CONH), 171.5 (CO₂CH₃). HRMS (CI⁺, CH₄) m/z: (MH⁺) calcd for C₁₆H₂₂NO₃, 276.1594; found, 276.1609. Anal. Calcd for C₁₆H₂₁NO₃: C, 69.79; H, 7.69; N, 5.09. Found: C, 69.31; H, 7.65; N, 5.15.

4.3.18.2. (S)-(Z)-Isomer 13f (less polar). (41 mg, 16%), as a white solid, mp 114–116 °C, $[\alpha]_D = +0.1$ (c 0.6, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 0.90 (d, J = 6.6 Hz, 3H, (CH₃)₂CH), 0.91 (d, J = 6.6 Hz, 3H, (CH₃)₂CH), 1.69 (m, 1H, CH(CH₃)₂), 2.51 (m, 1H, CH₂i-Pr), 2.58 (m, 1H, CH₂*i*-Pr), 3.73 (s, 3H, CO₂CH₃), 5.62 (d, J = 7.2 Hz, 1H, CHPh), 5.79 (dt, $J_{cis} = 11.6$, $J_{allylic} =$ 1.8 Hz, 1H, CHCO), 6.07 (dt, $J_{cis} = 11.6$, J = 7.2 Hz, 1H, CH*i*-Bu), 6.46 (d, J = 7.2 Hz, 1H, NH), 7.30–7.39 (m, 5H, H_{arom}). ¹³C NMR (100 MHz, CDCl₃): δ 22.5 ((CH₃)₂CH), 22.6 ((CH₃)₂CH), 28.8 (CH(CH₃)₂), 37.7 (CH₂*i*-Pr), 53.0 (CHPh), 56.4 (CO₂CH₃), 121.9, 127.4, 128.6, 129.1, 136.7, 146.4, 165.6 (CONH), 171.5 (CO₂CH₃). HRMS (CI⁺, CH₄) m/z: (MH⁺) calcd for C₁₆H₂₂NO₃, 276.1594; found, 276.1592. Anal. Calcd for C16H21NO3: C, 69.79; H, 7.69; N, 5.09. Found: C, 69.57; H, 7.62; N, 5.13.

4.3.19. (*S*)-*N*-(Cinnamoyl)-2-amino-2-phenylethanol 14a.²⁹ (302 mg, 89%), as a white solid, mp 197 °C, $[\alpha]_D = -29.8$ (*c* 2.47, MeOH). ¹H NMR (400 MHz, CD₃OD): δ 3.77 (dd, $J_{gem} = 11.4$, J = 8.0 Hz, 1H, CH₂OH), 3.80 (dd, $J_{gem} = 11.4$, J = 5.4 Hz, 1H, CH₂OH), 5.12 (dd, J = 7.2, J = 5.4 Hz, 1H, CHPh), 6.76 (d, $J_{trans} = 15.8$ Hz, 1H, CHCO), 7.24–7.41 (m, 8H, H_{arom}), 7.54 (d, $J_{trans} = 15.8$ Hz, 1H, CH=CHPh), 7.55–7.57 (m, 2H, H_{arom}). ¹³C NMR (100 MHz, CD₃OD): δ 57.4 (CH₂OH), 66.3 (CHPh), 121.8, 128.0, 128.4, 128.8, 129.5, 129.9, 130.8, 136.3, 141.2, 142.0, 168.3 (C=O). HRMS (CI⁺, CH₄) *m*/*z*: (MH⁺) calcd for C₁₇H₁₈NO₂, 268.1338; found, 268.1333. Anal. Calcd for C₁₇H₁₇NO₂: C, 76.38; H, 6.41; N, 5.24. Found: C, 76.04; H, 6.50; N, 5.33.

4.3.20. (*S*)-*N*-(*p*-Chlorocinnamoyl)-2-amino-2-phenylethanol **14b.**³⁰ (290 mg, 84%), as a white solid, mp 164 °C, $[\alpha]_{D} = -51.1$ (*c* 2.1, MeOH). ¹H NMR (400 MHz, CD₃OD): δ 3.78 (dd, $J_{gem} = 10.8$, J = 7.6 Hz, CH₂OH), 3.82 (dd, $J_{gem} = 10.8$, J = 5.2 Hz 1H, CH₂OH), 5.11 (dd, J = 7.6, J = 5.2 Hz, 1H, CHPh), 6.73 (d, $J_{trans} = 15.6$ Hz, 1H, CHCO), 7.23–7.39 (m, 7H, H_{arom}), 7.54–7.49 (d, $J_{trans} = 15.6$ Hz, 1H, CHAr), 7.52 (m, 2H, H_{arom}). ¹³C NMR (100 MHz, CD₃OD): δ 57.4 (CH₂OH), 66.3 (CHPh), 122.7, 128.0, 128.5, 129.5, 130.1, 130.3, 135.0, 136.5, 140.5, 141.1, 168.0 (C=O). HRMS (CI⁺, CH₄) *m*/*z*: (MH⁺) calcd for C₁₇H₁₇CINO₂, 302.0948; found, 302.0946 Anal. Calcd for C₁₇H₁₆CINO₂: C, 67.66; H, 5.34; N, 4.64. Found: C, 67.51; H, 5.43; N, 4.70.

4.3.21. (*S*)-*N*-(*p*-Methoxycinnamoyl)-2-amino-2-phenylethanol 14c. (328 mg, 87%), as a white solid, mp 158 °C, $[\alpha]_{D} = -71.4$ (*c* 2.6, MeOH). ¹H NMR (400 MHz, CD₃OD): δ 3.76 (dd, $J_{gem} = 11.2$, J = 7.6 Hz, 1H, CH₂OH), 3.80 (s, 3H, CH₃OPh), 3.81 (dd, $J_{gem} = 11.2$, J = 5.6 Hz, 1H, CH₂OH), 5.11 (dd, J = 7.6, J = 5.6 Hz, 1H, CH₂OH), 5.11 (dd, J = 7.6, J = 5.6 Hz, 1H, CH₂OH), 5.11 (dd, J = 7.6, J = 5.6 Hz, 1H, CHPh), 6.61 (d, $J_{trans} = 15.8$ Hz, 1H, CHCO), 6.93 (AA'BB' system, J = 8.8 Hz, 2H, H_{arom}), 7.23–7.38 (m, 5H, H_{arom}), 7.49 (d, $J_{trans} = 15.8$ Hz, 1H, CHAr), 7.50 (AA'BB' system, J = 8.8 Hz, 2H, H_{arom}). ¹³C NMR (100 MHz, CD₃OD): δ 55.9 (CH₂OH), 57.3 (CH₃OC₆H₄), 66.3 (CHPh), 115.4, 119.3, 128.1, 128.5, 128.9, 129.6, 130.6, 141.4, 141.9, 162.7, 168.9 (C=O). Anal. Calcd for

 $C_{18}H_{19}NO_3$: C, 72.71; H, 6.44; N, 4.71. Found: C, 72.85; H, 6.44; N, 4.71.

4.3.22. (*S*)-(*E*)-*N*-(3-tert-Butylacryloyl)-2-amino-2-phenylethanol 14d. (275 mg, 88%), as a white solid, mp 177 °C, $[\alpha]_{D} = +97.5$ (*c* 2.74, MeOH). ¹H NMR (400 MHz, CD₃OD): δ 1.08 (s, 9H, (CH₃)₃C), 3.72 (dd, $J_{gem} = 11.6$, J = 7.2 Hz, 1H, CH₂OH), 3.77 (dd, $J_{gem} = 11.6$, J = 5.6 Hz, 1H, CH₂OH), 5.05 (dd, J = 7.2, J = 5.6 Hz, 1H, CHPh), 5.99 (d, $J_{trans} = 15.6$ Hz, 1H, CHC=O), 6.80 (d, $J_{trans} = 15.6$ Hz, 1H, CHt-Bu), 7.24–7.34, (m, 5H, H_{arom}). ¹³C NMR (100 MHz, CD₃OD): δ 29.3 ((CH₃)₃CH), 34.3 (CH(CH₃)₃), 57.2 (CH₂OH), 66.3 (CHPh), 120.2, 128.1, 128.5, 129.6, 141.4, 155.9, 168.9 (C=O). Anal. Calcd for C₁₅H₂₁NO₂: C, 72.84; H, 8.56; N, 5.66. Found: C, 72.82; H, 8.71; N, 5.87.

4.3.23. (*S*)-(*E*)-*N*-(3-*iso*-Butylacryloyl)-2-amino-2-phenylethanol 14e. (254 mg, 81%), as a white solid, mp 117 °C, $[\alpha]_{D} = +66.2 (c \ 0.95, CHCl_3)$.¹H NMR (400 MHz, CDCl_3): δ 0.90 (d, J = 6.8 Hz, 6H, (CH₃)₂CH), 1.72 (m, 1H, CH(CH₃)₂), 2.03 (ddd, J = 6.8, $J_{allylic} = 1.2$ Hz, 2H, CH₂*i*-Pr), 3.54 (br, 1H, OH), 3.82 (s, 2H, CH₂OH), 5.05 (dd, J = 12.0, J = 5.2 Hz, 1H, CHPh), 5.84 (dd, $J_{trans} = 15.0$, $J_{allylic} = 1.2$ Hz, 2H, CHPh), 5.84 (dd, $J_{trans} = 15.0$, $J_{allylic} = 1.2$ Hz, 1H, CHC=O), 6.59 (br, 1H, NH), 6.82 (dt, $J_{trans} = 15.0$, J = 7.0 Hz, 1H, CH*i*-Bu), 7.25–7.35 (m, 5H, H_{arom}). ¹³C NMR (100 MHz, CDCl₃): δ 22.7 ((CH₃)₂CH), 22.7 ((CH₃)₂CH), 28.1 (CH(CH₃)₂), 41.6 (CH₂*i*-Pr), 56.2 (CH₂OH), 66.5 (CHPh), 124.2, 126.8, 127.9, 128.9, 139.1, 144.9, 166.6 (C=O). HRMS (FAB⁺) m/z: (MH⁺) calcd for C₁₅H₂₁NO₂: C, 72.84; H, 8.56; N, 5.66. Found: C, 72.51; H, 8.56; N, 5.78.

4.3.24. (*S*)-(*E*)-*N*-(3-*iso*-Propylacryloyl)-2-amino-2-phenylethanol 14f. (241 mg, 89%), as a white solid, mp 123–124 °C, $[\alpha]_D = +75.8$ (*c* 2.45, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ 1.03 (d, J = 6.8 Hz, 6H, (CH₃)₂CH), 2.42 (m, 1H, CH(CH₃)₂), 3.51 (br, 1H, OH), 3.83 (s, 2H, CH₂OH), 5.08 (dd, J = 12.4, J = 5.2 Hz, 1H, CHPh), 5.78 (dd, $J_{trans} = 15.6$, $J_{allylic} = 1.2$ Hz, 1H, CHC=O), 6.58 (d, J = 6.8 Hz, 1H, NH), 6.83 (dd, $J_{trans} = 15.6$, J = 6.6 Hz, 1H, CH*i*-Pr), 7.25-7.35 (m, 5H, H_{arom}). ¹³C NMR (100 MHz, CDCl₃): δ 21.6 ((CH₃)₂CH), 21.7 ((CH₃)₂CH), 31.0 (CH(CH₃)₂), 56.2 (CH₂OH), 66.6 (CHPh), 120.5, 126.8, 127.9, 128.9, 139.1, 152.1, 166.9 (C=O). HRMS (CI⁺, CH₄) m/z: (MH⁺) calcd for C₁₄H₂₀NO₂, 234.1494; found, 234.1504. Anal. Calcd for C₁₄H₁₉NO₂: C, 72.07; H, 8.21; N, 6.00. Found: C, 71.82; H, 8.21; N, 6.07.

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References

- (a) Tosaki, S. Y.; Tsuji, R.; Ohshima, T.; Shibasaki, M. J. Am. Chem. Soc. 2005, 127, 2147–2155; (b) Nemoto, T.; Kakei, H.; Gnanadesikan, V.; Tosaki, S. Y.; Ohshima, T.; Shibasaki, M. J. Am. Chem. Soc. 2002, 124, 14544–14545; (c) Caramella, P.; Reami, D.; Falzoni, M.; Quadrelli, P. Tetrahedron 1999, 55, 7027–7042; (d) Gothelf, R. F.; Jorgensen, K. A. Chem. Rev. 1998, 98, 863–909; (e) Jurczak, J.; Bauer, T.; Chapuis, C. In Stereoselective Synthesis (Houben-Weyl); Helchem, G., Hoffmann, R. W., Mulzer, J., Schaumann, E., Eds.; Thieme: Sttugart, 1995; Vol. E 21c, p 2735; (f) Cardillo, G.; Hashem, M. A.; Tomasini, C. J. Chem. Soc., Perkin Trans. 1 1990, 1587–1590.
- (a) Almaşi, D.; Alonso, D. A.; Najera, C. Tetrahedron: Asymmetry 2007, 18, 299–365; (b) Streuff, J.; Osterath, B.; Nieger, M.; Muñiz, K. Tetrahedron: Asymmetry 2005, 16, 3492–3496; (c) Cai, C.; Yamada, T.; Tiwari, R.; Hruby, V. J.; Soloshonok, V. Tetrahedron Lett. 2004, 45, 6855–6858; (d) Mpango, G. B.; Mahalanabis, K. K.; Mahdavi-Damghani, Z.; Snieckus, V. Tetrahedron Lett. 1980, 21, 4823–4826; (e) Mpango, G. B.; Snieckus, V. Tetrahedron Lett. 1980, 21, 4827–4830.
- (a) Sibi, M. P.; Chen, J.; Stanley, L. Synlett 2007, 298–302; (b) Evans, D. A.; Miller, S. J.; Lectka, T.; von Matt, P. J. Am. Chem. Soc 1999, 121, 7559–7573.
- (a) Amagata, T.; Minoura, K.; Numata, A. J. Nat. Prod. 2006, 69, 1384–1388; (b) Wu, Y. J.; He, H.; Sun, L. Q.; L'Heureux, A.; Chen, J.; Dextraze, P.; Starrett, J. E., Jr.; Boissard, C. G.; Gribkoff, V. K.; Natale, J.; Dworetzky, S. I. J. Med. Chem. 2004, 47, 2887–2896; (c) Wu, Y. J.; Davis, C. D.; Dworetzky, S.; Fitzpatricks, W. C.; Harden, D.; He, H.; Knox, R. J.; Newton, A. E.; Philip, T.; Polson, C.; Sivarao, D. V.; Sun, L. Q.; Tertyshnikova, S.; Weaver, D.; Yeola, S.; Zoeckler, M.; Sinz, M. W. J. Med. Chem. 2003, 46, 3778– 3781; (d) Cho, H.; Beale, J. M.; Graff, C.; Mocek, U.; Nakagawa, A.; Omura, S.; Floss, H. G. J. Am. Chem. Soc. 1993, 115, 12296–12304.
- 5. Bohlmann, F.; Gancer, M.; Kruger, M.; Nordhoff, E. *Tetrahedron* **1983**, *39*, 123–128.
- (a) Dias, L. C.; Melgar, G. Z.; Jardim, L. S. A. Tetrahedron Lett. 2005, 46, 4427-4431; (b) Diaper, C. M.; Sutherland, A.; Pillai, B.; James, M. N. G.; Semchuk, P.; Blanchard, J. S.; Vederas, J. C. Org. Biomol. Chem. 2005, 3, 4402-4411; (c) Ritter, T.; Kvaerno, L.; Werder, M.; Hauser, H.; Carreira, E. M. Org. Biomol. Chem. 2005, 3, 3514-3523; (d) Mattson, R. J.; Catt, J. D.; Denhart, D. J.; Deskus, J. A.; Ditta, J. L.; Higgins, M. A.; Marcin, L. R.; Sloan, C. P.; Beno, B. R.; Gao, Q.; Cunningham, M. A.; Mattson, G. K.; Molski, T. F.; Taber, M. T.; Lodge, N. L. J. Med. Chem. 2005, 48, 6023-6034; (e) Cecil, A. R. L.; Brown, R. C. D. Tetrahedron Lett. 2004, 45, 7269-7271; (f) Yokokawa, F.; Asano, T.; Okino, T.; Gerwick, W. H.; Shioiri, T. Tetrahedron 2004, 60, 6859-6880; (g) Crackett, P.; Demont, E.; Eatherton, A.; Frampton, C. S.; Gilbert, J.; Kahn, I.; Redshaw, S.; Watson, W. Synlett 2004, 679-683; (h) Xu, Z.; Chen, Z.; Ye, T. Tetrahedron: Asymmetry 2004, 15, 355-363; (i) Paintner, F. F.; Bauschke, G.; Polborn, K. Tetrahedron Lett. 2003, 44, 2549-2552; (j) Andrus, M. B.; Meredith, E. L.; Hicken, E. J.; Simmons, B. L.; Glancey, R. R.; Ma, W. J. Org. Chem. 2003, 68, 8162-8169; (k) Liu, F.; Zha, H.-Y.; Yao, Z.-J. J. Org. Chem. 2003, 68, 6679-6684; (1) Mulzer, J.; Riether, D. Org. Lett. 2000, 2, 3139-3141; (m) Martin, H. J.; Drescher, M.; Mulzer, J. Angew. Chem., Int. Ed. 2000, 39, 581-583; (n) Murakami, N.; Wang, W.; Ohyabu, N.; Ito, T.; Tamura, S.; Aoki, S.; Kobavashi, M.: Kitagawa, I. Tetrahedron 2000, 56, 9121-9128; (o) Rzasa, R. M.; Shea, H. A.; Romo, D. J. Am. Chem. Soc. 1998, 120, 591-592; (p) Romo, D.; Rzasa, R. M.; Shea,

H. A.; Park, K.; Langenhan, J. M.; Sun, L.; Akhiezer, A.; Liu, J. O. J. Am. Chem. Soc. **1998**, 120, 12237–12254; (q) Ahmar, M.; Duyck, C.; Fleming, I. J. Chem. Soc., Perkin Trans. 1 **1998**, 2721–2725; (r) Roush, W. R.; Reilly, M. L.; Koyama, K.; Brown, B. B. J. Org. Chem. **1997**, 62, 8708– 8721; (s) Norley, M.; Kocienski, P.; Faller, A. Synlett **1996**, 900–902; (t) Morimoto, Y.; Iwahashi, M.; Nishida, K.; Hayashi, Y.; Shirahama, H. Angew. Chem., Int. Ed. **1996**, 35, 904–906; (u) Kobayashi, M.; Wang, W.; Ohyabu, N.; Kurosu, M.; Kitagawa, I. Chem. Pharm. Bull. **1995**, 43, 1598– 1600; (v) Roush, W. R.; Brown, B. B. J. Org. Chem. **1993**, 58, 2162–2172; (w) Broka, C. A.; Ehrler, J. Tetrahedron Lett. **1991**, 32, 5907–5910.

- For reviews on the HWE reaction, see: (a) Maryanoff, B. E.; Reitz, A. B. Chem. Rev. **1989**, 89, 863–927; (b) Kelly, S. E. In Comprehensive Organic Synthesis; Trost, B. M., Fleming, I., Eds.; Pergamon Press: Oxford, 1991; Vol. 1, p 755; (c) Boutagy, J.; Thomas, R. Chem. Rev. **1974**, 74, 87–99; Papers relevant to this work: (d) Pedersen, T. M.; Hansen, E. L.; Kane, J.; Rein, T.; Helquist, P.; Norrby, P. O.; Tanner, D. J. Am. Chem. Soc. **2001**, *123*, 9738–9742; (e) Ando, K.; Oishi, T.; Hirama, M.; Ohno, H.; Ibuka, T. J. Org. Chem. **2000**, 65, 4745–4749; (f) Ando, K. J. Org. Chem. **1999**, 64, 6815–6821; (g) Brandt, P.; Norrby, P. O.; Martin, I.; Rein, T. J. Org. Chem. **1998**, 63, 1280–1289; (h) Ando, K. J. Org. Chem. **1997**, 62, 1934–1939.
- For asymmetric version of the HWE reaction, see: (a) Rein, T.; Pedersen, T. M. Synthesis 2002, 579–594; (b) Pedersen, T. M.; Jensen, J. F.; Humble, R. E.; Rein, T.; Tanner, D.; Bodmann, K.; Reiser, O. Org. Lett. 2000, 2, 535–538; (c) Rein, T.; Reiser, O. Acta Chem. Scand. 1996, 50, 369–379.
- Hernández-Fernández, E.; Fernández-Zertuche, M.; García-Barradas, O.; Muñoz-Muñiz, O.; Ordóñez, M. Synlett 2006, 440–444.
- For the use of chiral α,β-unsaturated amides bearing (S)-(αmethylbenzyl)benzylamine as a Michael acceptor, see: Castelot-Deliencourt, G.; Pannecoucke, X.; Quirion, J. C. Tetrahedron Lett. 2001, 42, 1025–1028.
- For the use of chiral α,β-unsaturated amides bearing (S,S)bis(α-methylbenzyl)amine as a building block, see: (a) Sato, M.; Gunji, Y.; Ikeno, T.; Yamada, T. Chem. Lett. 2004, 33, 1304–1305; (b) Riber, D.; Skrydstrup, T. Org. Lett. 2003, 5, 229–231; (c) Meth-Cohn, O.; Williams, D. J.; Chen, Y. Chem. Commun. 2000, 495–496; (d) Meth-Cohn, O.; Chen, Y. Tetrahedron Lett 1999, 40, 6069–6072; (e) Pearson, A. J.; Chang, K.; McConville, D. B.; Youngs, W. J. Organometallics 1994, 13, 4–5.
- For the use of chiral α,β-unsaturated amides bearing (S)-phenylglycine methyl ester as building block, see: (a) Ronald, G.; William, A. *Tetrahedron* 1988, 44, 1523–1534; (b) Norbert, E.; Boerries, K.; Wolfgang, S. Angew. Chem. 1977, 89, 408–410; For the other α-amino acids, see also: (c) Wijdeven, M. A.; Botman, P. N. M.; Wijtmans, R.; Schoemaker, H. E.; Rutjes, F. P. J. T.; Blaauw, R. H. Org. Lett. 2005, 7, 4005–4007; (d) Blaskovich, M. A.; Kahn, M. J. Org. Chem. 1998, 63, 1119–1125.
- For the use of chiral α,β-unsaturated amides bearing (S)phenylglycinol in asymmetric synthesis, see: (a) Tarver, J. E.; Terranova, K. M.; Joullié, M. M. Tetrahedron 2004, 60, 10277–10284; (b) Léautey, M.; Castelot-Deliencourt, G.; Jubault, P.; Pannecoucke, X.; Quirion, J. C. Tetrahedron Lett. 2002, 43, 9237–9240; (c) Castelot-Deliencourt, G.; Roger, E.; Pannecoucke, X.; Quirion, J. C. Eur. J. Org. Chem. 2001, 3031–3038; (d) Dahuron, N.; Langlois, N.; Chiaroni, A.; Richie, C. Heterocycles 1996, 42, 635–643; (e) Dahuron, N.; Langlois, N. Synlett 1996, 51–52; (f) Michelon, F.; Pouilhès, A.; Van Bac, N.; Langlois, N. Tetrahedron Lett. 1992, 33, 1743–1746; (g) Adam, W.; Güthlein, M.; Peters, K.;

Wirth, T. J. Am. Chem. Soc. **1998**, 120, 4091–4093; (h) Langlois, N.; Dahuron, N.; Wang, H. S. Tetrahedron **1996**, 25, 15117–15126.

- Ordóñez, M.; Hernández-Fernandez, E.; Xahuentitla, J.; Cativiela, C. Chem. Commun. 2005, 1336–1338.
- 15. Treatment of L-phenylglycinol with bromoacetyl bromide in the presence of K_2CO_3 in a CH_2Cl_2/H_2O mixture afforded the ester-amide derivative, which by Michaelis–Arbuzov reaction with trimethyl phosphite and subsequent hydrolysis using LiOH led to phosphonoacetamide 7 in 61% overall yield from L-phenylglycinol.
- 16. For the application of α-chloroamides bearing phenylalanine methyl ester in the preparation of β-lactams, see: Pérez-Faginas, P.; O'Reilly, F.; O'Byrne, A.; García-Aparicio, C.; Martínez-Martínez, M.; Pérez de Vega, M. J.; García-López, M. T.; González-Muñiz, R. Org. Lett. 2007, 9, 1593–1596.
- Bhattacharya, A. K.; Thyagarajan, G. Chem. Rev. 1981, 81, 415–430.
- For the biological evaluation of phosphonoamides, see: (a) Eldo, J.; Heng, S.; Kantrowitz, E. R. *Bioorg. Med. Chem. Lett.* 2007, 17, 2086–2090; (b) Gagnard, V.; Leydet, A.; Le Mellay, V.; Aubenque, M.; Morère, A.; Montero, J.-L. *Eur.* J. Med. Chem. 2003, 38, 883–891; (c) Kafarski, P.; Soroka, M. Synthesis 1982, 219–221; (d) Goodson, J. J.; Wharton, C. J.; Wrigglesworth, R. J. Chem. Soc., Perkin Trans. 1 1980, 2721–2727.
- (a) da Costa, J. C. S.; Pais, K. C.; Fernandes, E. L.; de Oliveira, P. S. M.; Mendonça, J. S.; de Souza, M. V. N.; Peralta, M. A.; Vasconcelos, T. R. A. *Arkivoc* 2006, 128–133; (b) Saeed, A.; Ashraf, Z. J. Chem. Sci. 2006, 118, 419–423.
- L-Phenylglycinol was readily obtained from reduction of Lphenylglycine with LiBH₄ in the presence of Me₃SiCl in dry THF at 0 °C, following the procedure described by: Nicolás, E.; Russell, K. C.; Hruby, V. J. J. Org. Chem. **1993**, 58, 766– 770.

- For the preparation of phosphonoamides from phosphonoacetic acid, see: (a) Grison, C.; Coutrot, P.; Comoy, C.; Balas, L.; Joliez, S.; Lavecchia, G.; Oliger, P.; Penverne, B.; Serre, V.; Hervé, G. Eur. J. Med. Chem. 2004, 39, 333–344; (b) Kummeter, M.; Kazmaier, U. Eur. J. Org. Chem. 2003, 3330– 3334; (c) Saltmarsh, J. R.; Boyd, A. E.; Rodriguez, O. P.; Radić, Z.; Taylor, P.; Thompson, C. M. Bioorg. Med. Chem. Lett. 2000, 10, 1523–1526; (d) Morris, A. D.; Cordi, A. A. Synth. Commun. 1997, 27, 1259–1266.
- Blanchette, M. A.; Choy, W.; Davis, J. T.; Essenfeld, A. P.; Masamune, S.; Roush, W. R.; Sakai, T. *Tetrahedron Lett.* 1984, 25, 2183–2186.
- For the HWE reaction of phosphonoamides using DBU as a base and additives, see: (a) Goodman, S. N.; Jacobsen, E. N. *Adv. Synth. Catal.* 2002, 344, 953–956; (b) Melekhov, A.; Fallis, A. G. *Tetrahedron Lett.* 1999, 40, 7867–7870; (c) Abiko, A.; Masamune, S. *Tetrahedron Lett.* 1996, 37, 1077–1080; (d) Oppolzer, W.; Dupuis, D.; Poli, G.; Raynham, T. M.; Bernardinelli, G. *Tetrahedron Lett.* 1988, 29, 5885–5888. Also, see: Refs. 6b,c,g.
- 24. When the minor isomers were not detected by ¹H NMR, a 98:02 *E:Z* ratio was established.
- 25. For HWE reaction at several temperatures, see: Sano, S.; Takemoto, Y.; Nagao, Y. Arkivoc 2003, 93–101.
- For HWE reaction of phosphonoamides with C₂ symmetry bearing (*R*,*R*)- and (*S*,*S*)-1,2-diphenyl-aziridine, see: Pedersen, T. B.; Jensen, K. F.; Humble, R. E.; Rein, T.; Tanner, D.; Bodmann, K.; Reiser, O. Org. Lett. 2000, 2, 535–538.
- For the HWE reaction of phosphonoamides bearing α-amino acids, see: Blaskovich, M. A.; Kahn, M. J. Org. Chem. 1998, 63, 1119–1125. Also, see: Refs. 6n,q,u.
- 28. Reyes, A.; Juaristi, E. Tetrahedron: Asymmetry 2000, 11, 1411–1423.
- 29. For the preparation of (R)-14a, see Ref. 13d.
- 30. For the preparation of (R)-14b, see Ref. 13h.